Nano-Solar cells: Solar Cells of the Future with Nanotechnology

Outline

1. Energy Research: Forefront and Challenges
   - Introduction – the energy challenge
   - Energy alternatives and the materials challenge
   - Think big, go small

2. Our Energy Future: Nano-Solar Cells

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Humanity’s Top Ten Problems for next 100 years

1. ENERGY
2. WATER
3. FOOD
4. ENVIRONMENT
5. POVERTY
6. TERRORISM & WAR
7. DISEASE
8. EDUCATION
9. DEMOCRACY
10. POPULATION

2007 6.6 Billion People
2050 8-10 Billion People


http://energysos.org/ricksmalley/top10problems/
“Tonight I’m proposing $1.2 billion in research funding so that America can lead the world in developing clean, hydrogen-powered automobiles... With a new national commitment, our scientists and engineers will overcome obstacles to taking these cars from laboratory to showroom, so that the first car driven by a child born today could be powered by hydrogen, and pollution-free.”

President Bush, State-of the-Union Address, January 28, 2003

“To finally spark the creation of a clean energy economy, we will double the production of alternative energy in the next three years.”

President Obama, George Mason University in Virginia, January 8, 2009
The World Energy Demand Challenge

World Fuel Mix 2001

- Oil
- Gas
- Coal
- Nuclear
- Renewables

World Energy Demand total
- Industrial
- Developing
- US
- EE/FSU

~ 14 TW by 2050
~ 33 TW by 2100

2100: 40-50 TW
2050: 25-30 TW
2000: 13 TW

Energy gap


EIA Intl Energy Outlook 2004
http://www.eia.doe.gov/oiaf/ieo/index.html

85% fossil
New Materials and Nanoscience will play a role in manipulating photons, electrons, and molecules.

- Manipulation of photons, electrons, and molecules
- Quantum dot solar cells
- Artificial photosynthesis
- Natural photosynthesis
- Nanostructured thermoelectrics
- Quantum dot solar cells
- TiO$_2$ nanocrystals
- Adsorbed quantum dots
- Liquid electrolyte

Nanoscale architectures:
- Top-down lithography
- Bottom-up self-assembly
- Multi-scale integration

Characterization:
- Scanning probes
- Electrons, neutrons, x-rays
- Smaller length and time scales

Theory and modeling:
- Multi-node computer clusters
- Density functional theory
- 10,000 atom assemblies

Solar energy requires interdisciplinary nanoscience research.
Solar Energy Utilization

**Solar Electric**
- 0.0002 TW PV (world)
- 0.00003 TW PV (US)
- $0.30/kWh w/o storage

1.5 TW electricity (world)
- $0.03-$0.06/kWh (fossil)

**Solar Fuel**
- 1.4 TW biomass (world)
- 0.2 TW biomass sustainable (world)

11 TW fossil fuel (present use)

**Solar Thermal**
- 0.006 TW (world)

- 500 - 3000 °C space, water heating
- 50 - 200 °C heat engines
- 500 - 3000 °C electricity generation
- process heat

~ 14 TW additional energy by 2050
What is a Solar Cell?

- It is also known as Photovoltaic cell (PV cell)
- A device that converts light energy (solar energy) directly to electricity.
- The term solar cell is designated to capture energy from sunlight, whereas PV cell is referred to an unspecified light source.
- It is like a battery because it supplies DC power.
- It is not like a battery because the voltage supplied by the cell changes with changes in the resistance of the load.
Applications of Solar Cells

- Renewable energy
- Can be powered for remote locations
- It’s free, limitless, and environmentally friendly...

- Toys, watches, calculators
- Electric fences
- Remote lighting systems
- Water pumping
- Water treatment
- Emergency power
- Portable power supplies
- Satellites
Moore’s Law for semiconductor electronics soon, all microchips will be nanoscale devices

CONCLUSION: Moore’s law continues for this decade regarding future size, device performance and cost for semiconductor electronics industry.

We now need to apply Moore’s law to set goals for the energy industry.
Example of Moore’s law

World PV Cell Production (MW)

Annual Growth > 30% For the Last Decade

Source: Paul Maycock, PV News, March 2006
Global Solar markets

Newly installed PV Power in 2007: 2.4 GWp

- Spain: 425 MWp; 18%
- Germany: 1100 MWp; 46%
- USA: 259 MWp; 11%
- Japan: 230 MWp; 10%
- Italy: 50 MWp; 2%
- Greece: 2 MWp; 0.1%
- France: 45 MWp; 2%
- Portugal: 10 MWp; 0%
- ROEU: 10 MWp; 0.4%
- China: 20 MWp; 1%
- South Korea: 50 MWp; 2%
- India: 20 MWp; 1%
- ROW: 175 MWp; 7%

Source: EPIA, ASIF, BSW
Updated February 2008
Reduction of environmental impact

Distribution of Renewable Energy Electricity Production in Germany 2007

- **Wind energy** 38.5 TWh; 48%
- **Hydro energy** 21.7 TWh; 27%
- **Bio energy solid** 6.6 TWh; 8%
- **Bio gas** 8.9 TWh; 11%
- **Bio energy liquid** 1.2 TWh; 2%
- **Geothermal** 0.1 TWh; 0.1%

Total RES electricity production 2007: 86.7 TWh
RES share of electricity consumption: 14.3%

Source: BEE, Jan 2008
Choice of solar technologies:

- Crystalline silicon
- Amorphous silicon
- Cadmium telluride
- Copper indium diselenide CIS family, notably copper indium gallium diselenide CIGS
- Dye sensitised solar cells DSSC
- Organic – polymer or small molecule
- Nano solar cell: silicon nanoparticle ink, carbon nanotube CNT and quantum dots, nanowires
- Inorganic nanorods embedded in semiconducting polymer, sandwiched between two electrodes
How Solar Cells Work: Photovoltaic Effect

We need to consider:

1. Energy source
2. Absorption
3. Transport
4. Collection

Entire spectrum of sunlight: 0.5 eV ~ 2.9 eV (Red light: 1.7 eV, Blue light: 2.7 eV)

- Photon excites valence electron
- Electron-hole pair created
- Electrons & holes “separate”
- Band gap determines what is absorbed
To free an electron, the energy of a photon must be at least as great as the bandgap energy.

\[ E_{\text{photon}} \geq E_{\text{band gap}} : \text{absorb to create free electrons.} \]
\[ E_{\text{photon}} < E_{\text{band gap}} : \text{pass through the material.} \]

\( E_{\text{band gap}} \) of other effective PV semiconductors ranges from 1.0 to 1.6 eV. In this range, electrons can be freed without creating extra heat.
Typical P-N Junction Silicon Solar Cell Structure

- Top side metallization grid
- Sunlight
- Anti-reflection coating
- n⁺ emitter
- p base
- Optional p⁺ back surface diffusion
- Back side metallization

Typical structure:
- 0.5 µm
- Average phosphorus ~ $10^{18}$ cm⁻³
- ~100-300 µm
- Boron ~ $10^{16}$ cm⁻³
Characterization of a Solar Cell Device

Photovoltaic power conversion efficiency of a solar cell

Current-voltage (I-V) curves of an organic solar cell (dark, - - -; illuminated, -). The characteristic intersections with the abscissa and ordinate are the open circuit voltage ($V_{oc}$) and the short circuit current ($I_{sc}$), respectively. The largest power output ($P_{max}$) is determined by the point where the product of voltage and current is maximized. Division of $P_{max}$ by the product of $I_{sc}$ and $V_{oc}$ yields the fill factor $FF$. $P_{in}$ is the incident light power density.

Generation photovoltaic

1\(^{st}\): Consists of a large-area, single layer p-n junction diode, which is capable of generating usable electrical energy from light sources with the wavelengths of sunlight. These cells are typically made using a silicon wafer.

2\(^{nd}\): Based on the use of thin-film deposits of semiconductors. These devices were initially designed to be high-efficiency, multiple junction photovoltaic cells.

3\(^{rd}\): Very different from the previous semiconductor devices as they do not rely on a traditional p-n junction to separate photogenerated charge carriers. These new devices include photoelectrochemical cells, polymer solar cells, and nanocrystal solar cells. Dye-sensitized solar cells are now in production. Examples include Amorphous silicon, Polycrystalline silicon, micro-crystalline silicon, Cadmium telluride, copper indium selenide/sulfide.

4\(^{th}\): Composite photovoltaic technology with the use of polymers with nano particles can be mixed together to make a single multispectrum layer. Then the thin multi spectrum layers can be stacked to make multispectrum solar cells more efficient and cheaper based on polymer solar cell and multi junction technology used by NASA on Mars missions.
## Comparison of thin film solar cells

<table>
<thead>
<tr>
<th>Type</th>
<th>Benefit or intended benefit</th>
<th>Efficiency</th>
<th>Challenges</th>
<th>Companies</th>
</tr>
</thead>
<tbody>
<tr>
<td>a-Si</td>
<td>Thin Film, Flexible substrates Potential for roll to roll processing unavailable for mono- or poly-Si. As above with high efficiency</td>
<td>8-10%</td>
<td>Constant degradation low efficiency Not tightlyrollable</td>
<td>Innovalight United Solar Mitsubishi Kovio</td>
</tr>
<tr>
<td>NanoSi</td>
<td></td>
<td>High?</td>
<td>Cd is toxic Controlled disposal only Not tightlyrollable</td>
<td>First Solar Calyxo</td>
</tr>
<tr>
<td>CdTe</td>
<td>Fairly high efficiency Well proven Over $1 billion of orders Lowest cost/watt over life</td>
<td>9-11%</td>
<td></td>
<td>HONDA Global Solar Wurth Nanosolar</td>
</tr>
<tr>
<td>CIGS</td>
<td>High efficiency at low cost long life transparent no disposal problems, printable</td>
<td>10-14%</td>
<td>Price of indium. New process/stability</td>
<td></td>
</tr>
<tr>
<td>DSSC</td>
<td>Tolerant of polarised/low level light-can use heat extreme angle of incoming light Transparent/colors, tightly rollable No disposal problems, printable</td>
<td>11%</td>
<td>Liquids handling Price of ruthenium? 5 year life?</td>
<td>G24i UK Dyesol Sony</td>
</tr>
<tr>
<td>Organic</td>
<td>Potential for lowest cost? Large area possible Tightly rollable No disposal problems, Spray directly onto things?</td>
<td>2-6%</td>
<td>Cost Efficiency 1 year life? Narrow spectrum</td>
<td>Konarka Plextronics Heliatek</td>
</tr>
</tbody>
</table>
Thin Film PV Technologies

- nanoparticle Si
- CIGS/CIS
- CdTe
- DSSC’s
- Organic PV
<table>
<thead>
<tr>
<th>Type of Cell/Technology</th>
<th>Best Power Conversion Efficiency (%)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Single crystalline silicon</td>
<td>24</td>
</tr>
<tr>
<td>Multi-crystalline silicon</td>
<td>18</td>
</tr>
<tr>
<td>Amorphous silicon</td>
<td>13</td>
</tr>
<tr>
<td>CIS, CIGS</td>
<td>~20</td>
</tr>
<tr>
<td>DSSC</td>
<td>11</td>
</tr>
<tr>
<td>Multi-Junctions (e.g. GaAs, InGaP)</td>
<td>~42</td>
</tr>
<tr>
<td>Organic &amp; hybrid solar cells</td>
<td>2-5</td>
</tr>
</tbody>
</table>

Source: NREL
Current Obstacles

Efficiency vs. cost

• Solar cell efficiencies vary from 6% for amorphous silicon-based solar cells to 42.8% with multiple-junction research lab cells.

• Solar cell energy conversion efficiencies for commercially available multicrystalline Si solar cells are around 14-16%.

• The highest efficiency cells have not always been the most economical — for example a 30% efficient multijunction cell based on exotic materials such as gallium arsenide or indium selenide and produced in low volume might well cost one hundred times as much as an 8% efficient amorphous silicon cell in mass production, while only delivering about four times the electrical power.
To Make an Efficient Solar Cell,

- Tune the p-layer to the properties of incoming photons to absorb as many as possible, and thus, to free up as many electrons as possible.
- Keep the electrons from meeting up with holes and recombining with them before they can escape from the PV cell.
- One designs the material to free the electrons as close to the junction as possible, so that the electric field can help send the free electrons through the conduction layer (the n-layer) and out into the electrical circuit.
- Antireflective Coating
  - Silicon reflects more than 30% of the light that shines on it.
  - To improve the conversion efficiency of a solar cell:
    1. Coat the top surface with a thin layer of silicon monoxide (SiO).
       a single layer: 10%; a second layer: less than 4%
    2. Texture the top surface: cones and pyramids which capture light rays
Revolutionary Photovoltaics: 50% Efficient Solar Cells

present technology: 32% limit for
- single junction
- one exciton per photon
- relaxation to band edge

multiple junctions
multiple gaps
multiple excitons per photon
hot carriers

rich variety of new physical phenomena
challenge: understand and implement
Thin Film Solar Cell Junction Structures

- **Homo junction**: p-n junction of crystalline silicon

- **Hetero junction**: formed by contacting two different semiconductors—CdS and CuInSe₂

- **p-i-n / n-i-p**: typically, amorphous silicon thin-film cells use a p-i-n structure, whereas CdTe cells use an n-i-p structure.

- **Multi junction**: also called a cascade or tandem cell, can achieve a higher total conversion efficiency by capturing a larger portion of the solar spectrum

A multijunction device is a stack of individual single-junction cells in descending order of bandgap ($E_g$).

The top cell captures the high-energy photons and passes the rest of the photons on to be absorbed by lower-bandgap cells.
Prospects for Nano-enhanced Solar Cells (Ying Guo)

Basic research underway with the technology developments required to achieve the desired applications.
Solar Cells based on Nanotechnology:

Organic and Hybrid cells
Nano-crystalline TiO$_2$ Film: Dye-Sensitized Solar Cell (DSSC)
Quantum dot solar cells
Nanowire solar cells
Organic and hybrid solar cells

Why organic solar cells?

ADVANTAGES

• Easier and cheaper fabrication processes: non-vacuum, low temperature, available at industrial scale.
• Low cost materials (?), low quantity (g/m²), flexible and cheap substrates (no glass).
• High EQE over the whole sun spectrum.
• High light absorption (100nm are enough to absorb most of the light)

DISADVANTAGES

• Lower carrier mobility than in inorganic semiconductors.
• New devices, a lot of work for optimising morphology and composition.
Examples of organic semiconductors used in organic solar cells

Organic Hetero-Junction Cell

- The first cells were built in the 50s (organic dyes with inorganic semiconductors).
- First organic cells had very low efficiency: $10^{-2}/10^{-3}$%.
- Due to the very short diffusion length for excitons (about 10nm), only the charges very close to the junction could be efficiently separated and collected by the electrodes.

Illustration of the photoinduced charge transfer with a sketch of the energy level. After excitation in the PPV polymer, the electron is transferred to the C60...
The Bulk Hetero-Junction

• A composite of two organic semiconductor (n and p) sandwiched between two electrodes with work function matching the energy bands of the composite.
• It can be seen as nanoscale p-n junctions.
• The density of p-n junctions is much higher than in bilayer cells.
• Much larger interfacial area, where charges can be separated.
• Efficiency increased of 2-3 order of magnitude reaching 1%.
• Separate conduction path for electron and holes (like in DSC).
Organic cells can be a cheap alternative to inorganic semiconductor cells: low cost material, cheap fabrication process, good spectral response.

What has to be done?

- Optimising the morphology is the key for better efficiencies.
- Two competing goals: good mixing, partial ordering.
Hybrid cells

BHJ with organic/inorganic semiconductors

ex. P3OT:CdTe, P3HT:CdSe, CuPC:Si, P3HT:TiO₂, P3HT:CIS

Maximum reported eff. is 2%

ADVANTAGES

Inorganic semiconductors have much higher carrier mobility.

Inorganic semiconductor can be produced in the form of nanocrystal in order to control the ordering of the microstructure.
Ideal structure of a bulk heterojunction solar cell

Interspaced with an average length scale: 10-20 nm ≤ Exciton diffusion length

The two phases have to be interdigitated in percolated highways to ensure high mobility charge carrier transport with reduced recombination.

Buried Nanoelectrodes

Separate the absorption process from the charge transport in case of different mobility for holes and e\(^{-}\) in the blend.
Hybrid solar cells with vertically aligned CdTe nanorods and a conjugated polymer
Yoonmook Kang, Nam-Gyu Park, and Donghwan Kimb

Fig. 2. Scanning electron microscopy images: (a) the top view of the vertical aligned CdTe nanorods; (b) the cross section of the composite layer.
The adsorbed dye molecule absorbs a photon forming an excited state. \([\text{dye}^*]\)
The excited state of the dye can be thought of as an electron-hole pair (exciton).

The excited dye transfers an electron to the semiconducting TiO\(_2\) (electron injection).
This separates the electron-hole pair leaving the hole on the dye. \([\text{dye}^{**+}]\)

The hole is filled by an electron from an iodide ion.
\([2\text{dye}^{**+} + 3\text{I}^- \rightarrow 2\text{dye} + \text{I}_3^-]\)
Nano-crystalline TiO2 Film
Dye-Sensitized Solar Cell (DSSC)

- Electrons are collected from the TiO2 at the cathode.
- Anode is covered with carbon catalyst and injects electrons into the cell regenerating the iodide.
- Redox mediator is iodide/triiodide (I-/I3⁻)
- The dashed line shows that some electrons are transferred from the TiO2 to the triiodide and generate iodide. This reaction is an internal short circuit that decreases the efficiency of the cell.
Solid-state dye-sensitized solar cell

Light absorption in dye, electron transfer to TiO₂, hole transfer to Spiro-MeOTAD.

Additives in HTM: tbp and Lithium TFSI

Application of Carbon Nanotubes to Counter Electrodes of Dye-sensitized Solar Cells


Table 1. Conversion efficiency ($\eta$) of the cells and specific surface area ($S$) of carbon materials

<table>
<thead>
<tr>
<th>Catalyst</th>
<th>$\eta$/%</th>
<th>$S$/m$^2$ g$^{-1}$</th>
</tr>
</thead>
<tbody>
<tr>
<td>SWCNT</td>
<td>3.5</td>
<td>764</td>
</tr>
<tr>
<td>Carbon filament</td>
<td>2.5</td>
<td>350</td>
</tr>
<tr>
<td>Nanohorn</td>
<td>2.4</td>
<td>300</td>
</tr>
<tr>
<td>Pt</td>
<td>5.4</td>
<td>—</td>
</tr>
<tr>
<td>None</td>
<td>0.1</td>
<td>—</td>
</tr>
</tbody>
</table>

Table 2. Conversion efficiency ($\eta$) of the cells and sheet resistance ($R$) of carbon electrodes

<table>
<thead>
<tr>
<th>Electrode</th>
<th>$\eta$/%</th>
<th>$R$/Ω square$^{-1}$</th>
</tr>
</thead>
<tbody>
<tr>
<td>SWCNT</td>
<td>4.5</td>
<td>1.8</td>
</tr>
<tr>
<td>Carbon filament</td>
<td>0.2</td>
<td>990</td>
</tr>
<tr>
<td>Nanohorn</td>
<td>0.04</td>
<td>&gt;1000</td>
</tr>
</tbody>
</table>
Key Step – Charge Separation

- Charge must be rapidly separated to prevent back reaction.
- Dye sensitized solar cell, the excited dye transfers an electron to the TiO₂ and a hole to the electrolyte.
- In the PN junction in Si solar cell has a built-in electric field that tears apart the electron-hole pair formed when a photon is absorbed in the junction.

Basic problems needed to overcome:

1. the high recombination rate at the TiO₂ interface
2. the low conductivity of the hole conductor itself.
Solar Cells based on Nanotechnology:

Quantum dot solar cells
Nanowire solar cells
Strategies for Improvement
(with nanostructures- QDs)

- Capture more sunlight - Tune energy band gaps of materials

  “Stacked Architecture”

  Potential Advantages:
  - Cheaper Materials
  - Capture more of Solar Spectrum

- Generate more electrons per photon:

  “Multiple Exciton Generation”

  Potential Advantages:
  - Dramatic Efficiency Improvements
  - Greater Solar Spectrum
MEG- Multiple Exciton Generation

Basic Solar Cell:

1 photon = 1 exciton
(1 electron/1 hole + excess energy)

MEG Solar Cell:

1 photon = 2+ excitons
(2+ electron/holes –reduce heat loss!)

Potential Efficiency Improvement
Why Quantum Dots for solar cells?

Thermal relaxation of excited charge carriers can be significantly slowed down.

1. Enhanced photovoltage = collect charges while their hot.
2. Enhanced photocurrent = get more from the hot ones.


Detection

- Higher photon energy/band gap ratios give higher carrier multiplication efficiencies.
- Onset at $\sim 3E_g$.


The Trick:

- Have to be able to extract charge carriers produced in quantum dots.

The two fundamental pathways for enhancing the conversion efficiency (increased photovoltage [7,8] or increased photocurrent [9,10] can be accessed, in principle, in three different QD solar cell configurations; these configurations are shown in Fig. 2 and they are described below. However, it is emphasized that these potential high-efficiency configurations are speculative and there is no experimental evidence yet that demonstrates actual enhanced conversion efficiencies in any of these systems.
Coaxial silicon nanowires as solar cells and nanoelectronic power sources

Device fabrication and diode characterization

Characterization of the p-i-n silicon nanowire photovoltaic device

The overall apparent efficiency of the p-i-n coaxial silicon nanowire photovoltaic elements—3.4% (upper bound) and 2.3% (lower bound)—exceeds reported nanorod/polymer and nanorod/dye systems, and could be increased substantially with improvements in Voc by means of, for example, surface passivation.

Application of the p-i-n silicon nanowire photovoltaic device

Figure 4 | Self-powered nanosystems. a, Real-time detection of the voltage drop across an aminopropyltriethoxysilane-modified silicon nanowire at different pH values. The silicon nanowire pH sensor is powered by a single silicon nanowire photovoltaic device operating under 8-sun illumination ($V_{oc} = 0.34$ V, $I_{sc} = 8.75$ nA). Inset, circuit schematics. b, Light $I$–$V$ curves (1-sun, AM 1.5G) of two silicon nanowire photovoltaic devices (PV 1 and PV 2) individually and connected in series and in parallel. c, Nanowire AND logic gate powered by two silicon nanowire photovoltaic devices in series. Insets, circuit schematics and truth table for the AND gate. The resistance of CdSe nanowire is $\sim 5 \ \text{G}\Omega$; the $V_{oc}$ of two photovoltaic devices in series is 0.53 V. The large resistance of the CdSe nanowire and reverse-biased p-i-n diode makes $V_a$ and $V_i$ (HIGH) very close to $V_{oc}$ of the photovoltaic device. To get $V_i$ (LOW), the diode is simply grounded.

Efficient charge separation for solar cells could be achieved by applying strain to silicon nanowires.

The added strain in the nanowire changes the structure of energy bands, such that positive and negative charges can be separated into different regions of the nanowire. The main potential benefit of silicon nanowire cells over traditional crystalline silicon cells is that the silicon would not need to be doped with other materials. This means that nanowire solar cells could be made inexpensively from much lower quality silicon, and without labour-intensive processing.

Nanowire applications for solar cells

Nanowires decouple light absorption and carrier extraction into different directions!

Atwater, Caltech
Choice of solar technologies:

- Crystalline silicon
- Amorphous silicon
- Cadmium telluride
- Copper indium diselenide CIS family, notably copper indium gallium diselenide CIGS
- Dye sensitised solar cells DSSC
- Organic – polymer or small molecule
- Others such as silicon nanoparticle ink, carbon nanotube CNT and quantum dots
- Nano solar cell: Inorganic nanorods embedded in semiconducting polymer, sandwiched between two electrodes
Summary

- A mix of future sustainable energy conversion technologies will be needed.
- New materials and nanoscience discoveries are necessary to its development.
- Strong interplay between basic and applied sciences is a key to success.
- Interdisciplinary approaches, and coupling theory/experiment are vital.
- Working with industry at all stages is a key factor.
- The challenges and constraints are global and complementary among different countries.
- International collaboration and networking must be encouraged and supported.